

1 – PRODUCT AND COMPANY IDENTIFICATION

PRODUCT NAME:	. VAT BOOSTER
CHEMICAL NAME/	
CLASS/SYNONYMS:	Caustic Soda 50% Solution
PRODUCT NUMBER:	. VAT BOOSTER
UN/NA NUMBER:	. 1824
CHEMICAL FAMILY:	. Sodium salt
CAS NUMBER:	• Not applicable for mixtures.
FORMULA:	Mixture
COMPANY:	. JMN Specialties, Inc.
COMPANY:	. JMN Specialties, Inc. 1100 Victory Drive – Westwego, Louisiana USA 70094
COMPANY:	• /
COMPANY:	1100 Victory Drive – Westwego, Louisiana USA 70094
	1100 Victory Drive – Westwego, Louisiana USA 70094 Phone (504) 341-3749, Fax (504) 341-5868
	1100 Victory Drive – Westwego, Louisiana USA 70094 Phone (504) 341-3749, Fax (504) 341-5868 www.jmnspecialties.com
	 1100 Victory Drive – Westwego, Louisiana USA 70094 Phone (504) 341-3749, Fax (504) 341-5868 www.jmnspecialties.com CALL CHEMTEL: Toll Free US & Canada: (800) 255-3924, Outside USA +01-813-248-0585.

2 – HAZARDS IDENTIFICATION

GHS HAZARD CLASSIFICATION:

Physical Hazards

Flammable Liquids:..... No hazard statement

Health Hazards

WARNING LABEL ITEMS INCLUDING PRECAUTIONARY STATEMENTS:



SIGNAL WORD:..... DANGER!

GHS HAZARD AND PRECAUTIONARY STATEMENTS:

H290: May be corrosive to metals H301 H311 H331: Toxic if swallowed, in contact with skin or if inhaled H314: Causes severe skin burns and eye damage H318: Causes serious eye damage

P101+102+103: If medical advice is needed, have product container or label at hand. Keep out of the reach of children. Read label before use.

P202+270+280+281: Do not handle until all safety precautions have been read and understood. Do not eat, drink or smoke when using this product. Wear protective gloves/protective clothing/eye protection/face protection. Use personal protective equipment as required.



P301+330+331: IF SWALLOWED: Rinse mouth. Do NOT induce vomiting P303+361+353: IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Rinse skin with water/shower P304+341: IF INHALED: If breathing is difficult, remove victim to fresh air and keep at rest in a position comfortable for breathing P305+351+338: IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses if present and easy to do – continue rinsing

P363: Wash contaminated clothing before reuse

P405: Store locked up

P406: Store in a corrosive resistant container with a resistant inner liner

P501: Dispose of contents/container: Treatment, storage, transportation and disposal must be in accordance with Federal, State/Provincial and Local Regulations, and product characteristics at time of disposal.

TOTAL VOC's: None

3 – COMPOSITION / INFORMATION ON INGREDIENTS			
HAZARDOUS INGREDIENT	PERCENT	CAS NUMBER	
Sodium Hydroxide	49 - 51	1310-73-2	
Water	49 - 51	7732-18-5	

4 - FIRST-AID MEASURES

BREATHING (INHALATION):	Remove from exposure area to fresh air immediately. If breathing has stopped, perform artificial resuscitation. Keep person warm and at rest. Treat symptomatically and supportively. Seek medical attention immediately. Qualified medical personnel should consider administering oxygen.
SWALLOWING (INGESTION)	: Give large amounts of fresh water or milk immediately. Do not give anything by mouth if person is unconscious or otherwise unable to swallow. If vomiting occurs, keep head below hips to prevent aspiration. Treat symptomatically and supportively. Seek medical attention immediately.
EYES:	. Flush eye with copious quantities of water. If persistent irritation
	 occurs, obtain medical attention. Remove contaminated clothing and wash affected skin with soap and water. If persistent irritation occurs, obtain medical attention. When using high pressure equipment, injection of product under the skin can occur. If high pressure injuries occur, the casualty should be sent immediately to a hospital. Do not wait for symptoms to develop. May cause caustic burns to the mouth, throat or stomach if swallowed. After swallowing danger of stomach perforation. On inhalation: Irritation of mucous membrane, coughing and shortness of breath. All treatments should be based on observed signs and symptoms of distress in the patient. Contact poison treatment specialist immediately if large quantities have been ingested or inhaled. Consideration should be given to the possibility that overexposure to materials other than this



5 – FIRE-FIGHTING MEASURES

GENERAL FIRE HAZARDS:	 Fire fighters should wear full protective clothing, including self- contained breathing equipment.
AUTOIGNITION TEMP:	. NA
EXTINGUISHING MEDIA:	Determined by surrounding material. In case of fire, use water fog, dry chemical, CO_2 , or "alcohol" foam. Firefighting should be attempted only by those who are adequately trained and equipped with proper protective equipment.
SPECIAL FIRE FIGHTING	
PROCEDURES:	• No action shall be taken involving any personal risk or without suitable training. Evacuate surrounding areas. Keep unnecessary and unprotected personnel from entering. Do not touch or walk through spilled material. Provide adequate ventilation. Wear appropriate respirator when ventilation is inadequate. Put on appropriate personal protective equipment. Spilled product may be slippery.
UNUSUAL FIRE AND	
EXPLOSION HAZARDS:	• Containers may explode from internal pressure if confined to fire. Cool with water spray.

6 – ACCIDENTAL RELEASE MEASURES

SPILL PROCEDURES:	 Wear appropriate personal protective equipment before approaching spill site. For small spills, dilute with water to sewer if allowed by local and state regulations. If unable to wash product with water, absorb with inert material (sand or other approved material) and dispose of in accordance with applicable regulations. Treatment, storage, transportation and disposal must be in accordance with Federal, State/Provincial and Local Regulations. Regulations may vary in different locations. Characterization and compliance with applicable laws are the responsibility solely of the generator. Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste disposal facility. Processing, use or contamination of this product may change the waste management
RCRA STATUS:	 options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in accordance with federal, state and local requirements. Sodium Hydroxide (Caustic Soda), if discarded or spilled, as well as other wastes generated during use of sodium hydroxide or containing sodium hydroxide may exhibit one or more hazardous waste characteristics under 40 CFR 261.24: D002 – Corrosive.

7 – HANDLING and STORAGE

STORAGE:Keep in a tightly closed container, stored in a cool, dry, ventilated area
below 44°C (110°F). Protect against physical damage. Isolate from
incompatible substances. Containers of this material may be hazardous
when empty since they retain product residues (vapors, liquid); observe
all warnings and precautions listed for the product. Drum must not be
washed out or used for other purposes.



HANDLING:...... Avoid contact with eyes, skin and clothing. Do not inhale vapors and fumes. Wash thoroughly after handling. Use only with adequate ventilation. Do not take internally. For industrial use only.

8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

OCCUPATIONAL EXPOSURE LIMITS

EXPOSURE CONTROLS:

HAZARDOUS	INGREDIENT
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Sodium Hydroxide

Water

 2 mg/m^3

PEL

TLV-TWA 2 mg/m³

None Established

None Established



Good general ventilation (typically 10 air changes per hour) should be used. Ventilation rates should be matched to conditions. If applicable, use process enclosures, local exhaust ventilation, or other engineering controls to maintain airborne levels below recommended exposure limits. If exposure limits have not been established, maintain airborne levels to an acceptable level. Please refer to the ACGIH document, *Industrial Ventilation, A Manual of Recommended Practices*, most recent edition, for details.

RESPIRATORY PROTECTION: If engineering controls do not maintain airborne concentrations below recommended exposure limits (where applicable) or to an acceptable level (in countries where exposure limits have not been established), an approved respirator must be worn. In the United States of America, if respirators are used, a program should be instituted to assure compliance with OSHA Standard 63 FR 1152, January 8, 1998. Respirator type: Air-purifying respirator with an appropriate, government approved (where applicable), air-purifying filter, cartridge or canister. Contact health and safety professional or manufacturer for specific information.
 PROTECTIVE CLOTHING: Eye/face protection: Wear chemical goggles; face shield (if splashing is possible). Skin protection: Chemical resistant, impermeable gloves.

EVECTOTION: Every acceptotection: wear chemical goggles; face sheld (if splashing is possible). Skin protection: Chemical resistant, impermeable gloves. Gloves should be tested to determine suitability for prolonged contact. Use of impervious apron or chemical suit and chemical resistant boots are recommended.

ADDITIONAL MEASURES: Handle in accordance with good industrial hygiene and safety practice. Wash thoroughly with soap and water after handling and before eating, drinking, or using tobacco. Safety shower and eye wash should be available close to work areas.



9 – PHYSICAL / CHEMICAL PROPERITES

FLASHPOINT: Non-flammable material UPPER FLAME LIMIT (%): NA LOWER FLAME LIMIT (%): ... NA **VAPOR DENSITY (AIR=1):.....** 1.38 - (Air = 1.0) SPECIFIC GRAVITY: 1.525 SOLUBILITY IN WATER:..... Complete VOLATILITY INCLUDING WATER: 12.72 pounds per gallon MOLECULAR WEIGHT: 40.1 g/mol EVAPORATION RATE: NA PHYSICAL STATE: Liquid COLOR: Clear ODOR:.....Bland

10 – STABILITY and REACTIVITY

STABILITY: Stable

HAZARDOUS DECOMP.:..... Will not occur

INCOMPATIBILITY:	Avoid direct contact with water and strong acids. Add slowly to water
	or acids with dilution and agitation to avoid a violent exothermic or
	explosive reaction. Avoid contact with aluminum, tin, zinc, leather, and
	organic halogen or nitro compounds. Sodium hydroxide in contact with
	acids and organic halogen compounds, especially trichloroethylene,
	may causes violent reactions. Contact with nitromethane and other
	similar nitro compounds causes formation of shock-sensitive salts.
	Contact with metals such as aluminum, magnesium, tin, and zinc cause
	formation of flammable hydrogen gas. Sodium hydroxide, even in fairly
	dilute solution, reacts readily with various sugars to produce carbon
	monoxide. Precautions should be taken including monitoring the tank
	atmosphere for carbon monoxide to ensure safety of personnel before
	vessel entry.
HAZARDOUS REACTIONS:	Contact with metal may release flammable hydrogen gas. Reacts
	violently with strong acids. This product may react with oxidizing
	agents. Do not mix with other chemicals.

11 – TOXICOLOGICAL INFORMATION

THRESHOLD LIMIT VALUE:.. 2 mg/m³

 OSHA PEL: 2 mg/m³

 LISTED CARCINOGEN: This product IS NOT listed in the National Toxicology Program (NTP) Report on Carcinogens (latest edition) or has been found to be a potential carcinogen in the International Agency for Research on Cancer (IARC) Monographs (latest editions) or found to be a potential carcinogen by OSHA.



MEDICAL CONDITION

AGGRAVATED: Pre-existing medical conditions of the following organ(s) or organ system(s) may be aggravated by exposure to this material: Respiratory system. Eyes. Skin.

INFORMATION ON ACUTE TOXICOLOGICAL EFFECTS

ORAL

a burning sensation in the mouth, corrosion of thelips, mouth, tongue and pharynx, and severe esophageal and abdominal pain, vomiting of blood and large pieces of mucosa, and bloody diarrhea. Asphyxia can occur from swelling of the throat. Mediastinitis, alkalemia, pallor, weak, slow pulse, cardiovascular collapse, shock, coma and death may occur. Perforation of the alimentary tract and constrictive scarring may result. Esophageal stricture may occur weeks, months, or even years later to make swallowing difficult. The estimated fatal dose in man is 5grams. Cases of squamous cell carcinoma of the esophagus have occurred with latent periods of 12 to 42 years after ingestion. These cancers were believed to be sequela of tissue destruction and possibly scar formation rather than the result of direct carcinogenic action of sodium hydroxide.

DERMAL

skin fissures and white eschars may occur without immediate pain. Exposure to solutions as weak as 0.03 N (0.12%) for 1 hour has caused injury to healthy skin. With solutions of 0.4-4%, irritation does not occur until after several hours. Solutions of 25-50% caused no sensation of irritation within 3 minutes in human subjects. Skin biopsies from human subjects having 1 N sodium hydroxide applied to their arms for 15 to 180 minutes showed progressive changes beginning with dissolution of the cells in the horny layer and progressing through edema to total destruction of the epidermis in 60 minutes. A 5% aqueous solution caused severe necrosis to the skin of rabbits when applied for 4 hours. Alkalies penetrate the skin slowly. The extent of injury depends on the duration of contact. If sodium hydroxide is not removed from the skin, severe burn swith deep ulceration may occur. Exposure to the dust or mist may cause multiple small burns and temporary loss of hair. Pathologic findings due to alkalies may include gelatinous, necrotic areas at the site of contact.

INHALATION

of the nose at 2 mg/m^3 to severe pneumonitis depending on the severity of exposure. Low concentrations may cause mucous membrane irritation with sore throat, coughing, and dyspnea. Intense exposures may result in destruction of mucous membranes and delayed pulmonary edema or pneumonitis. Shock may occur.

REPEATED DOSE TOXICITY

dusts for 30 years or more found no significant increase in mortality in relation to duration or intensity of such exposures.

SKIN CORROSION / IRRITATION

Product: Effects are dependent upon concentration and duration of exposure. Dermatitis or effects similar to those for acute exposure may occur.

SERIOUS EYE DAMAGE / IRRITATION

corneal epithelium, corneal opacification, marked edemaand ulceration. After 7 to 13 days either gradual recovery begins or there is progression of ulceration and corneal opacification. Complications of severe eye burns are symblepharon with overgrowth of the cornea by a vascularized membrane, progressive or recurrent corneal ulceration and permanent corneal opacification. Blindness may occur.

RESPIRATORY OR SKIN SENSITIZATION

dusts for 30 years or more found no significant increase in mortality in relation to duration or intensity of such exposures.



MUTAGENCITY

IN VITRO Product: No Data Available IN VIVO Product:..... No Data Available **Specified Substance(s)**

Sodium Hydroxide 50% Solution

Information as provided by manufacturer

No Data Available

CARCINOGENICITY

industrial experience with caustic soda, there is no evidence that caustic is a skin sensitizer or is readily absorbed through the skin. It is not a known carcinogen, mutagen, developmental toxicant or reproductive toxicant. This product does not contain any chemicals known to State of California to cause cancer, birth defects, or any other reproductive harm.

REPODUCTIVE TOXICITY

classified as hazardous.

SPECIFIC TARGET ORGAN TOXICITY - SINGLE EXPOSURE

of the respiratory tract. Severe exposures could result in chemical pneumonia. EYES: Contact can cause severe damage including burns and blindness. The severity of the effects depend on concentration and how soon after exposure the eyes are washed. SKIN: Corrosive. Contact may cause burns and tissue destruction. Note that irritation may follow an initial latency (delay between the time the exposure occurs and when the sense of irritation starts). The latent period can vary as much as hours for a dilute solution (0.04%) to minutes with more concentrated solutions (25 - 50%). Prolonged or repeated contact, even to dilute concentrations, can cause a high degree of tissue destruction. INGESTION: Corrosive. Severe burns and complete tissue perforation of mucous membranes of mouth, throat and stomach.

SPECIFIC TARGET ORGAN TOXICITY – REPEATED EXPOSURE

Product: CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS: Not available. TERATOGENIC EFFECTS: Not available. DEVELOPMENTAL

TOXICITY: Not available. The substance is toxic to lungs. Repeated or prolonged exposure to the substance can produce target organs damage. Repeated or prolonged contact with spray mist may produce chronic eye irritation and severe skin irritation. Repeated or prolonged exposure to spray mist may produce respiratory tract irritation leading to frequent attacks of bronchial infection. Repeated exposure to a highly toxic material may produce general deterioration of health by an accumulation in one or many human organs.

ASPIRATION HAZARD

vomiting may cause chemical pneumonia.

OTHER ADVERSE EFFECTS

Product: No data available

12 – ECOLOGICAL INFORMATION

ACUTE TOXICITY

FISH

Product: Bluegill sunfish: 48-hour LC50 = 99 mg/L Mosquito fish: 96-hour LC50 = 125 mg/L Brown shrimp (Crangon crangon): 48-hour LC50 = 30 - 100 mg/L



AQUATIC INVERTEBRATES

CHRONIC TOXICITY

FISH

AQUATIC INVERTEBRATES

Product:......Expected to have low toxicity: 10 < LC/EC/IC50 <= 100 mg/l TOXICITY TO AQUATIC PLANTS

Product: Freshwater algae are destroyed above pH 8.5.

PERSISTENCE AND DEGRADABILITY

BIODEGRADATION

BIOLOGICAL OXYGEN DEMAND

Product: No data available

CHEMICAL OXYGEN DEMAND

Product: No data available

BOD / COD RATIO

Product: No data available

BIOACCUMULATIVE POTENTIAL

MOBILITY IN SOIL

Product: Expected to partition to water. The pH effect of sodium hydroxide in water is naturally reduced by the absorption of atmospheric carbon dioxide. This reduction is also effected by dilution with water and by the natural acidity of a given water body. There is no degradation of sodium hydroxide in waters, only loss by absorption or through chemical neutralization.

RESULTS OF PBT AND mPvB ASSESSMENT

OTHER ADVERSE EFFECTS

13 – DISPOSAL CONSIDERATIONS

WASTE DISPOSAL: Treatment, storage, transportation and disposal must be in accordance with Federal, State/Provincial and Local Regulations. Regulations may vary in different locations. Characterization and compliance with applicable laws are the responsibility solely of the generator. Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste disposal facility. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in accordance with federal, state and local requirements.



other wastes generated during use of sodium hydroxide or containing sodium hydroxide may exhibit one or more hazardous waste characteristics under 40 CFR 261.24: D002 - Corrosive.

14 – TRANSPORTATION INFORMATION

Important Note: Shipping descriptions may vary based on mode of transport, quantities, package size, and/or origin and destination. Consult your company's Hazardous Materials/Dangerous Goods expert for information specific to your situation.



UN/NA NUMBER:	. 1824
PROPER SHIPPING NAME:	Sodium Hydroxide, solution
HAZARD CLASS:	.8
PACKAGING GROUP :	. II
LETTER:	C (Corrosive substances)
ENVIRONMENTAL HAZARD:	Environmental Hazard Value Score (IRCH) = 29. Caustic soda does not
	bioaccumulate due to its high solubility in water. It is considered
	slightly toxic to aquatic organisms unless there is a significant pH shift
	outside the range of $5 - 10$, which may be toxic to aquatic organisms.
REPORTABLE QUANTITY:	(RQ) (40 CFR 302.4): Sodium Hydroxide 50% Solution CAS# 1310-
-	73-2, 2,000 lb.

15 - REGULATIONS

This Safety Data Sheet conforms to ANSI Z400.5, and to the format requirements and the International Chemical Safety Cards of the Global Harmonizing System. This SDS complies with 29 CFR 1910.1200 (HAZARD COMMUNICATION STANDARD). IMPORTANT: Read this SDS before handling & disposing of this product. Pass this information on to employees, customers, & users of this product.

EPA SRA Title III Chemical Listings:

	• This product is listed on the TSCA inventory. If this product is a blend, all ingredients in the product are listed on the TSCA Inventory List. Any impurities present in this product are exempt from listing.
SECTION 302:	
SECTION 304:	. None
SECTION 312:	. THRESHOLD PLANNING QUANTITY (40 CFR 370): The TPQ
	for Sodium Hydroxide 50% Solution (Caustic Soda 50% Solution)
	CAS# 1310-73-2 is 2,000 lbs.
SARA SECTION 313:	. SECTION 313 REPORTABLE INGREDIENTS (40 CFR 372):
	This product contains toxic chemicals subject to the reporting requirements of Section 313, Title III of the SARA (Superfund
	Amendments and Reauthorization Act) of 1986: Sodium Hydroxide
	(Caustic Soda)
ACUTE:	. Yes
CHRONIC:	. Yes
FIRE:	. No
PRESSURE:	• No

PAGE 9 of 10



IMDG – International Marine Dangerous Goods Code

UN1824, Sodium Hydroxide, Solution, 8, PG II. EmS F-A, S-B. Marine Pollutant: No. **IATA** UN1824, Sodium Hydroxide, Solution, 8, PG II.

DEA Chemical Trafficking Act:.. No

16 – OTHER INFORMATION

HMIS*

IIIIIB		
HEALTH		3
FLAMMABILITY		0
REACTIVITY		1
PERSONAL PROTECTI	ON	х

***HMIS®HAZARD INDEX: 0=Minimal Hazard, 1=Slight Hazard, 2=Moderate Hazard, 3=Serious Hazard, 4=Severe Hazard.** HMIS® rating involves data interpretations that may vary from company to company. They are intended only for rapid, general identification of the magnitude of the specific hazard. To deal adequately with the safe handling of this material, all the information contained in this SDS and product label must be considered.

ND = No Data, NA = Not Applicable/Not Available, \leq = Less than or equal to, \geq = Greater than or equal to

REVISION STATEMENT: Changes have been made throughout this Safety Data Sheet (SDS). Please read the entire document. Prepared according to the OSHA Hazard Communication Standard (29 CFR 1910.1200) and The Globally Harmonized System of Classification and Labeling of Chemicals (GHS) by the Company Health and Risk Assessment Unit.

DISCLAIMER:

Although the information and recommendations set forth herein (hereinafter "Information") are presented in good faith and believed to be correct as of the date hereof, the Company makes no representations as to the completeness or accuracy thereof. Information is supplied upon the condition that the persons receiving this Safety Data Sheet (SDS) will make their own determination as to its suitability for their intended purposes prior to use. Since the product is within the exclusive control of the user, it is the user's obligation to determine the conditions of safe use of this product. Such conditions should comply with all Federal and State Regulations concerning the Product. It must be recognized that the physical and chemical properties of any product may not be fully understood and that new, possibly hazardous products may arise from reactions between chemicals. The information given in this data sheet is based on our present knowledge and shall not constitute a guarantee for any specific product features and shall not establish a legally valid contractual relationship. NO REPRESENTATIONS OR WARRANTIES, EITHER **EXPRESS** OR IMPLIED, OF MERCHANTABILITY, FITNESS FOR A PARTICULAR PURPOSE OR OF ANY OTHER NATURE ARE MADE HEREUNDER WITH RESPECT TO INFORMATION OR THE PRODUCT TO WHICH **INFORMATION REFERS.**

This is the last page of this SDS